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Carbonate alkalinity is determined by titration of the water sample to the phenolphthalein indicator endpoint, or approximately a pH of 8.3. Total alkalinity is determined by titration of the water sample to the endpoint of the methyl orange indicator, or an approximate pH of 4.5.

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in fresh water,  $I=0.002$ ,  $\gamma_{O_2} = 1.0006$  Molar vs. Molal in principle, activity predictions are based on molal concentrations (mole/kg solvent), but since we are often most concerned with dilute solutions, we frequently use molar concentrations. David Reckhow CEE 680 #4 24

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Alkalinity - Whitman College

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