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A secondary standard
solution is a chemical
term that refers to a

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solution that has its concentration measured by titration with a primary standard solution, explains EasyChem.com. The amount of chemical reactants in the primary standard solution is known beforehand.

What Is a Secondary Standard Solution? | Reference.com

There are two types of

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standard solutions known as primary solution and secondary solution. A primary standard solution is a solution with a high purity and less reactivity. A secondary standard is not that pure and is chemically reactive than primary standards.

IUPAC - standard solution (S05924)

A secondary standard is a chemical that has

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been standardized against a primary standard for use in a specific analysis. Secondary standards are commonly used to calibrate analytical methods. NaOH, once its concentration has been validated through the use of a primary standard, is often used as a secondary standard.

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A solution of accurately known concentration, prepared using standard substances in one of several ways. A primary standard is a substance of known high purity which may be dissolved in a

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known volume of solvent to give a primary standard solution. If @S06026@ is used to establish the strength of a @T06386@, it is called a secondary standard ...

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Solutions - 6 - My Brother Sam is Dead Literature Guide My Brother Sam is Dead Sample Agenda Our Literature Guides are designed to be used in their sequential entirety, or may be divided into separate parts. Not all activities must be used, but to achieve full comprehension and mastery of the

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Difference Between Primary and Secondary Standard Solution ...

The key difference
between primary and
secondary standard
solution is that primary

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standard solution has a high purity and less reactivity whereas secondary solution has a less purity and high reactivity.

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