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Organic compounds are distinct from other materials in that they are mainly made of hydrogen, carbon, nitrogen,

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oxygen and smaller amounts of a few other elements. They make up most of the mass in living things. In this introductory non-lab course, you'll learn the role that organic compounds play in nature, as well as their industrial applications.

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Isomerism is especially important in organic chemistry because of the

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capacity of carbon atoms to be arranged in so many different ways: continuous chains, branched chains, and rings. Structural formulas can be written so that every bond is shown, or in various abbreviated forms. For example, the formula for n-pentane (n stands for

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