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CHAPTER 1 \_\_\_\_ 1. (a)

90.50% carbon; 9.50%  
hydrogen (b) C<sub>4</sub>H<sub>5</sub> 2.

32.0% carbon; 5.4%  
hydrogen; 62.8%

chlorine; C<sub>3</sub>H<sub>6</sub>Cl<sub>2</sub> 3.

C<sub>2</sub>H<sub>5</sub>NO<sub>2</sub> 4. 180.2 =  
molecular mass.

Molecular formula is C

9H<sub>8</sub>O<sub>4</sub> 5.

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molecular mass.  
Molecular formula is  
C<sub>9</sub>H<sub>8</sub>O<sub>4</sub>. 5. Equivalent  
weight = 52.3 6. (a) 7.

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In simple words, it is a  
method to measure  
how much light is  
absorbed by a  
chemical substance  
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of light passes through  
it.

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