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active as a consultant to the College Board for AP Chemistry and is a member of the exam task force for the Chemistry Olympiad program of the American Chemical Society.

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the 2018 Administration Author: The
College Board Subject: AP Chemistry
Scoring Guidelines - 2018 Keywords: AP
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The molality of the solution is the moles of solute per kilogram of solvent. Since water has a density of one kilogram per liter, we can simply divide the total moles by the liters of solution. The molality of the solution is . We can treat this solution as 3L of water containing

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4mol of a solute that dissolves into a total of 5 ions.

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Questions page 4 1994 Discuss the following phenomena in terms of the chemical and physical properties of the substances involved and general principles of chemical and physical change. (a) As the system shown above approaches equilibrium, what change occurs to the volume of water in

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