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your notebook, answer the following SECTION 7.1 IONS For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most

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(the answer is not an electron configuration, even though it is) ...

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the 2. that matter
undergoes. ... Answer
the following questions
in the space provided.

28. Match each activity
below to one of the five
branches of chemistry.

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Chemistry [Antony C. Wilbraham] on Amazon.com. *FREE* shipping on qualifying offers. # This Resource package includes: # Planning Guides, Practice Problems, Section Reviews, Solutions Manual

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Valence and core

Electrons Name Date O

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O Purpose To discover the arrangements of electrons within atoms. Complete the table on the handout, filling in the missing atoms.

Then answer the questions. I. How does the number of electrons change as you move from left to right a period? ... Living By Chemistry Teaching and Classroom Resources @ 2012 W

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Choose the best answer and write its letter on the line. 1. Which of the following tasks would probably not be assigned to an ... Discuss the role of chemistry in one of these areas: Materials, Energy, Medicine and

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- Choose the best answer and write its letter on the line.
11. The fourth principal energy level has a. 4 orbitals. b. 16 orbitals. c. 32 orbitals. d. 9 orbitals.
12. If the electron configuration of an element is $1s^2 2s^2 2p^6 3s^2 3p^5$, the element is a. iron. b. bromine. c. chlorine. d. phosphorus.

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