

## Chemistry Chapter 5 Review Answers

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AP Chemistry Review Questions - Gases and Gas Laws ... 5.65 mol ? 37.0 mol; A 268 cm<sup>3</sup> sample of an ideal gas at 18°C and 748 Torr pressure is placed in an evacuated container of volume 648cm<sup>3</sup>. To what centigrade temperature must the assembly be heated so that the gas will fill the whole chamber at 748 Torr? ?

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*Unit 2 - Chapters 4, 5 & 6 - Mrs. Gingras' Chemistry Page*  
CHAPTER 6 REVIEW Chemical Bonding SECTION 5 SHORT ANSWER Answer the following questions in the space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear ...

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Mrs. Gingras' Chemistry Page. Home Chemistry 111/112 Chemistry 122 About Contact ... Chapters 4, 5 & 6. This unit's focus is on atomic structure. Students will cover concepts related to atomic mass, isotopes, electron configurations and periodicity. ... Chapter 4 Review Answers: File Size: 141 kb: File Type: doc: Download File. Chapter 4 Quiz ...

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Identify the letter of the choice that best completes the statement or answers the question. \_\_\_\_ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to. a. Mendeleev. c. Bohr. ... Chemistry Chapter 5 Exam ...

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CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy required to remove an electron from a neutral ...

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Write empirical formulas to match the following molecular formulas: CH<sub>3</sub>CO<sub>2</sub>H a. C<sub>2</sub>H<sub>6</sub>O<sub>4</sub>N<sub>2</sub>O<sub>5</sub> b. N<sub>2</sub>O<sub>5</sub>HgCl c. Hg<sub>2</sub>Cl<sub>2</sub>CH<sub>2</sub> d. C<sub>6</sub>H<sub>12</sub>O<sub>2</sub> C<sub>4</sub>H<sub>8</sub>A certain hydrocarbon has an empirical formula of CH<sub>2</sub> and ...

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