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The result is the Sweet  
16 Chemistry  
Compound  
Tournament. Hopefully,  
your students will  
enjoy playing the  
“Tournament” while  
reviewing these  
important topics. The  
rules for filling out the  
Tournament brackets  
are simple: For the first  
round, the student  
determines the  
chemical formula for  
the compound and  
counts the total



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Chemistry  
number of atoms in ...

Formulas

## **Sweet 16 Chemistry Tournament Compound**

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the corner, your  
students' thoughts will

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soon be turning to  
sunshine, prom, and  
the NCAA basketball  
tournament. This  
clever activity  
combines the ever-  
popular March  
Madness basketball  
pool with a review of  
chemical formulas, the  
structure and charges  
of ions, and molar  
mass calculations. The  
result is the Sweet 16  
Chemistry Compound  
Tournament.

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## Chemistry **Sweet 16 Chemistry Compound**

### **Tournament - Flinn**

Two versions of the Sweet 16 Chemistry Compound Tournament are enclosed. One uses traditional chemical names [e.g. ferrous, cupric, etc] and the other uses the Stock system for chemical names [iron(II), copper(II)].

## **Answer Key Sweet 16 Chemistry**

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## Chemistry Tournament **Compound Tournament**

The result is the Sweet  
16 Chemistry  
Compound

Tournament. Hopefully,  
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important topics. The  
rules for filling out the  
Tournament brackets  
are simple: For the first  
round, the student  
determines the  
chemical formula

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## **Sweet 16 Chemistry Compound Tournament**

Sweet 16 Chemistry of  
Gases Tournament?

First round: Predict the  
Gaseous product.

Second round: Predict  
the one that dissolves  
in water to give an  
acidic solution. Semis:  
The gas that is more...

## **Sweet 16 Chemistry of Gases Tournament? |**



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Compound Tournament  
Ammonium sulfate.

Potassium

permanganate.

Calcium hypochlorite.

Potassium ferricyanid

e. Mercury(I) sulfate.

Titanium(IV) oxid e.

Iron(III) sulfate. Sodium

...

**Sweet 16 Chemistry  
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**Flinn Sweet Sixteen**  
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## Chemistry Tournaments **Ion Tournament Solubility PDF**

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Answers  
spring just around the  
corner, your students'  
thoughts will soon be  
turning to sunshine,  
prom, and the NCAA  
basketball tournament.  
This clever activity  
combines the ever-  
popular basketball  
office pool with a  
review of the formulas  
of ionic com-pounds  
and their solubility.

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chemistry! . The  
compound with a  
polyatomic ion is  
declared the winner!

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## **Sweet 16 Chemical Formulas Tournament - Flinn**

...

Sweet 16 Chemistry  
Ion Tournament First  
Round Second Round  
Semifinals Finals  
Winner NiCl<sub>2</sub> Ni(II)<sub>2</sub>+  
Fe(II) S Zn CO<sub>3</sub> Na  
CrO<sub>4</sub> Mg PO<sub>4</sub> Ag(I) SO<sub>4</sub>  
Ba NO<sub>3</sub> Al OH Rules 1.  
First Round →Add

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Charges 2. Second  
Round → Write Formula  
3. Semis → Soluble  
Compound (compound  
that will dissolve in  
water) 4. Finals  
→ Soluble Product of  
Chemical Reaction 5.

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