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correct answer. ii) $k = 2.5 \times 10^{-2} \text{ mol}^{-1} \text{ s}^{-1}$
 $10 \text{ [R]} \text{ [S]}$ rate $u = 1.1 \times 10^{-4} \text{ mol}^{-1} \text{ s}^{-1}$
Another example of rearranging the
formula before you can put the numbers
in, one mark is given for the correct

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formula full marks for the correct answer. 1 iii) T 1 Hint: If you calculated a value for $k > 4.2 \times 10^{-4}$ then you will be allowed the answer to iii ...

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Chapter 3 2 a) $K_a = \frac{[CH_3COO^-][H^+]}{[CH_3COOH]}$
1.35 10^{-5} 0.169 candidates
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round brackets are used, this will be
penalised but examiners will continue to
mark the rest of the question without
further penalty.

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Chapter 8 c) i) X 1 ii) Y 1 7 Structure 1 =
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gained by realising that an amide is
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Chapter 2 Answers 1 a) $K_c = \frac{\text{Marks [H}_2\text{] }^3}{\text{[C}_2\text{H}_2\text{]}^3} = 1$ If round brackets are used, this will be penalised but examiners will continue to mark the rest of the question without further penalty.

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