

## Answers To Chemquest 30

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90 ChemQuest 30 Name ... Compare your answers to questions 2 and 4. They are both approximately the same. Therefore, it takes about 6.02x10<sup>23</sup> atoms of carbon to equal the atomic mass in grams and it takes 6.02x10<sup>23</sup> atoms of phosphorus to ... 1 ChemQuest 1 Name: \_\_\_\_ Page 3 and 4: 3 Another example: 7.45 ...

**90 ChemQuest 30 Name:**  
Chemquest\_intro\_to\_moles\_answer - ChemQuest 30.I.I.I mane-c Date.--L New iaLaLs Hour Information Atomic Mass Using Atomic Mass Units(amu and Grams)9

**Chemquest Intro to Moles answer - ChemQuest 30.I.I.I mane ...**  
Ch. 8: ChemQuest #30 (Intro to Moles), ChemQuest#31a, #31b (% composition Divide moles of reactants by the balancing coefficient; smallest answer is LR. When all of themembers of your group can get the answer to #7 and 8 by showing the right setup, then stapie your 6. if calcium atomsweight 40. g/mole, then how much would 3.0 moles weigh?

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