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16—3 Review and Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperature affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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Study Guide and Reinforcement 3 ANSWER KEY 7. opposes the motion of objects that move through the air, is affected by speed, size, and shape 8. net force 9. microwelds 10. rolling 11. air resistance 12. acceleration 13. sliding 14. parachute 1. Gravity is a force that every object in the

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11—1 Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles of magnesium are required to react with 2.0 mol of hydrochloric acid? The equation for this reaction is  $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$  14. Aluminum reacts with HCl to produce aluminum chloride ( $\text{AlCl}_3$ ) and hydrogen ...

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2 Answer.Key 2-1 Explore, page 5 The original source of energy for this activity would be the stored chemical energy in the dry cell battery. The chemical energy in the battery is converted to electrical energy, which is changed to mechanical energy in the running of the motor. Finally, the mechanical

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3/13/98 14 - 1 14. Using Computational Aerodynamics: Review and Reinforcement It is a well-known phenomena that computer programs, left to themselves, will cease to work. This is known as "Bit Rot". Programs NOT left to themselves also cease to work. This is known as "Enhancement". The FLOPS Motto, Arnie McCullers, ViGYAN, Inc.

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Gas Stoichiometry will help to determine volume ratios for gaseous reactants and products by using coefficients from a chemical equation. Coefficients in balanced equations represent the number of moles and liters.

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